

## Chapter 15

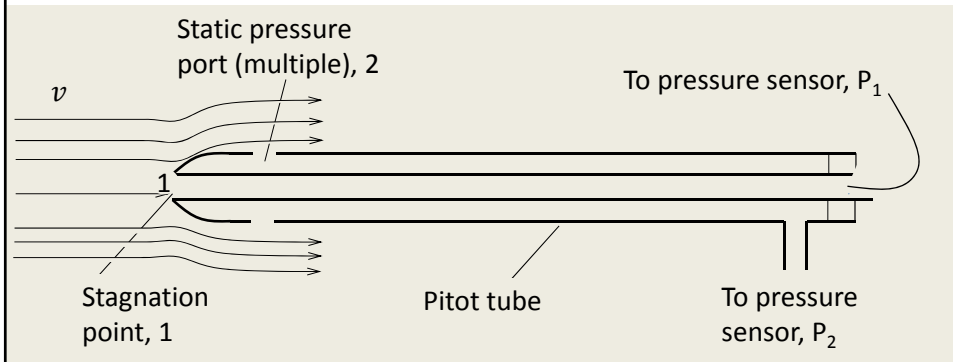
### **Air Monitoring**

#### Source Sampling

- A representative sample is the key to the accurate measurement of emissions from a source.
- Different source test methods have been developed for a variety of realistic environments.
- In the USA, for example, most states rely upon either US EPA Reference Methods, Methods 1 to 8, for characterization of the gas flow and specific air pollutants from a stationary source.

### Flow Rate and Velocity Measurement

- There are a number of flow rate measurement technologies and related instruments; one of them is Pitot tube
- It is widely employed in engineering practices because of its simplicity, accuracy, reliability and cost-effectiveness.
- A standard Pitot tube is an L-shaped tube
- When it is located in a pipe, streamlines connect the stagnation point and the multiple static taps, which are small holes evenly distributed along the circumference of the outer tube.



$$\Delta P = P_1 - P_2 = \frac{1}{2} \rho u_2^2$$

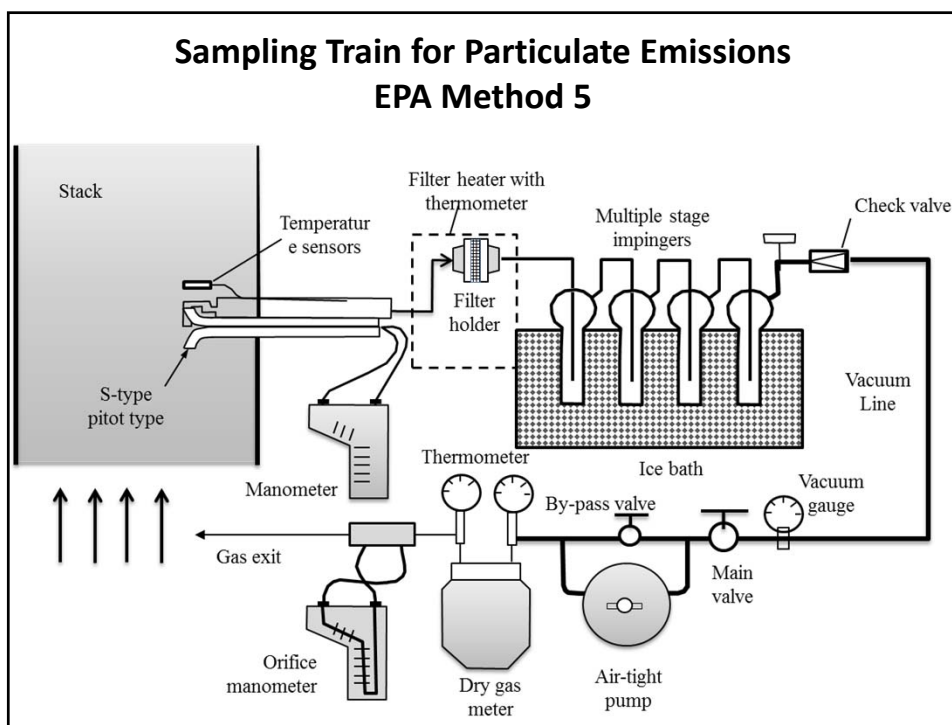
$$u = u_2 = \left( \frac{2\Delta P}{\rho} \right)^{1/2}$$

$$Q = Au$$

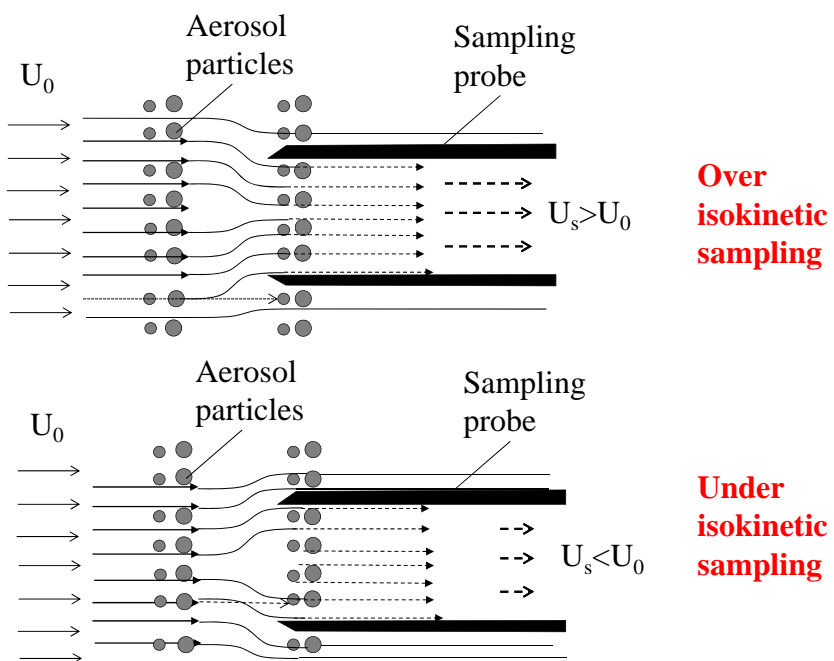
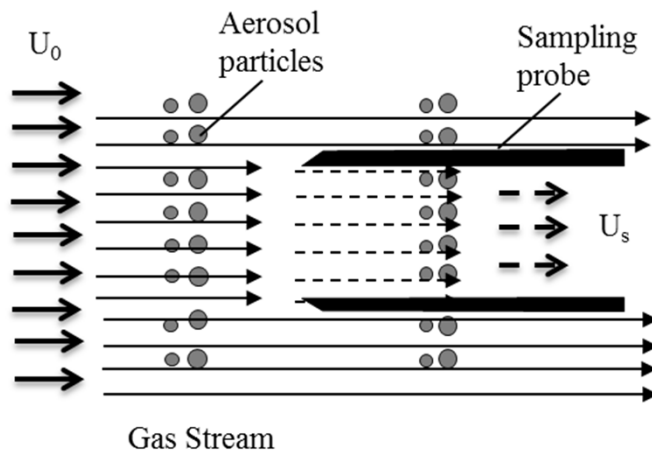
## Source Sampling

### US EPA Methods 1-8

- Method 1: Sample and velocity transverse for stationary sources
- Method 2: Determination of stack gas velocity and volumetric flow rates using Type S Pitot tube,
- Method 3: Gas analysis for the determination of dry molecular weight
- Method 4: Characterizing moisture content in stack gases
- Method 5: Determination of particulate matter emission from stationary sources
- Method 6: Determination of sulfur dioxide emissions from stationary sources
- Method 7: Determination of nitrogen oxide emissions from stationary sources
- Method 8: Determination of sulfuric acids and sulfur dioxide emissions from stationary sources



## Isokinetic sampling $U_s = U_0$



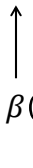
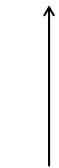
## Sampling efficiency $\eta_s = C_s/C_o$

Sampling efficiency  $\eta_s$  due to velocity mismatch only

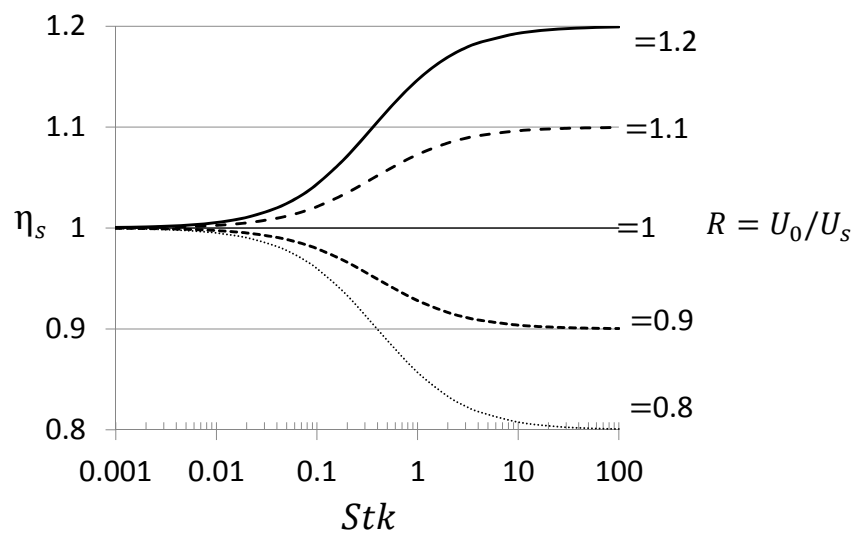
$$\eta_s = 1 + (R - 1)\beta(Stk, R)$$

$$\beta(Stk, R) = 1 - \frac{1}{1 + (2 + 0.617/R)Stk}$$

$$R = U_0/U_s$$



Sampling efficiency  $\eta_s$  vs.  $R$  and  $Stk$   
(Velocity mismatch only)



## Sampling Efficiency due to Misalignment Effect

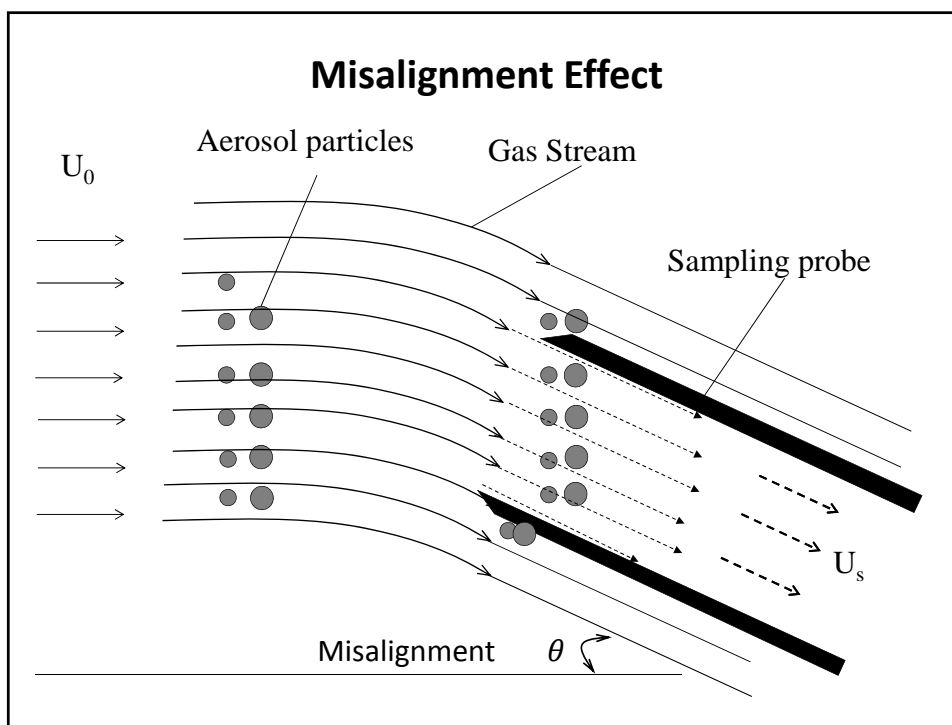
- Empirical equations were given by [Duraham and Lundgren \(1980\)](#) to quantify the misalignment effect for anisokinetic sampling as follows.

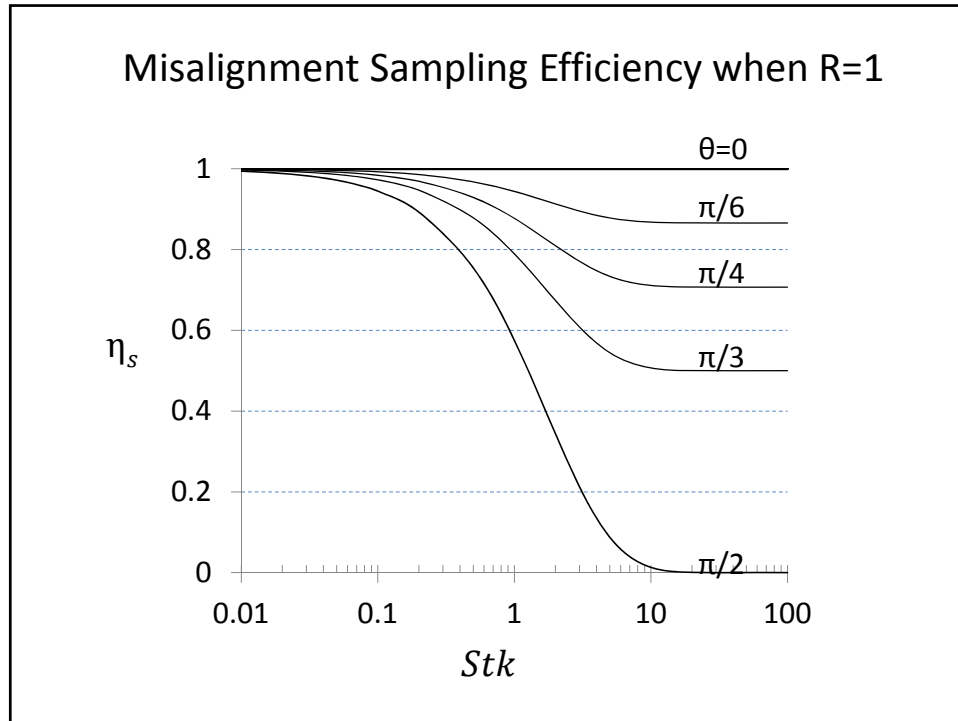
$$\eta_s = 1 + (\cos\theta - 1)\beta'(Stk', \theta) \quad \text{for } R = 1$$

$$\eta_s = 1 + (R\cos\theta - 1)\beta'(Stk', \theta) \frac{\beta(Stk', R)}{\beta(Stk', R = 1)} \quad \text{for } R \neq 1$$

$$\beta'(Stk', \theta) = 1 - \frac{1}{1 + 0.55Stk' \exp(0.25Stk')}$$

$$Stk' = Stk \times \exp(0.022\theta) \quad \text{for } 0 \leq \theta < \pi/2$$

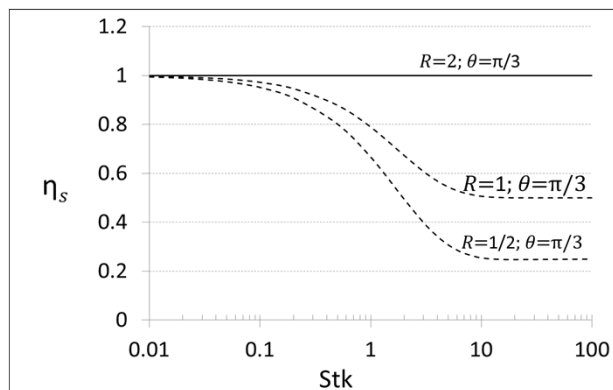




### Misalignment and $R \neq 1$

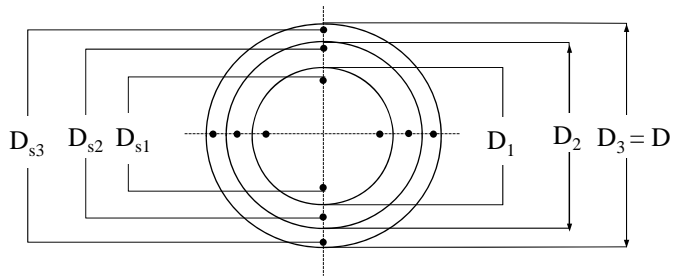
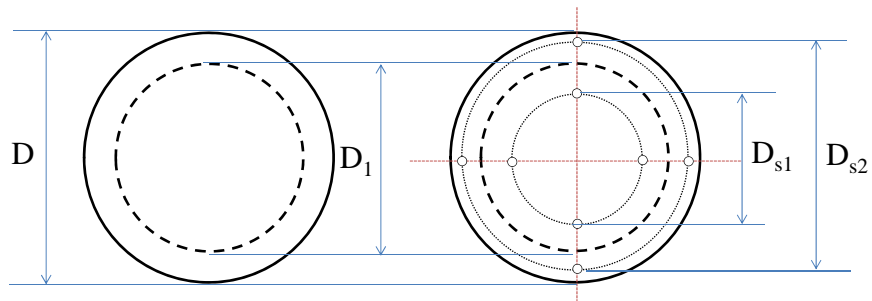
$$\eta_s = 1 + (R \cos \theta - 1) \beta'(Stk', \theta) \frac{\beta(Stk', R)}{\beta(Stk', R = 1)}$$

- However, it is useful to realize that the sampling efficiency is always 1 when  $R \cos \theta = 1$



## Multiple Point Source Sampling

- Equal area rule



$$D_{si}^2 = \frac{i}{2n} D^2$$

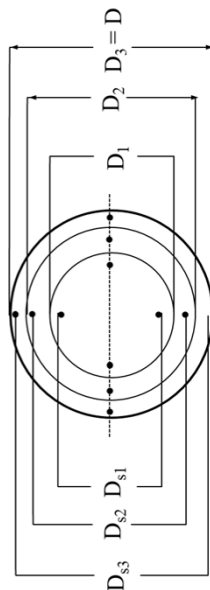
$$D_{si} = D \sqrt{\frac{i}{2n}} \text{ where } i = 1, 3 \dots 2n - 1$$

$$D_i = D \sqrt{\frac{i}{2n}} \text{ where } i = 2, 4 \dots 2n$$



**Example  
15.1**

A stack with an inner diameter of 1.2 m is subject to source tests. Determine the sampling points by dividing the stack area into 3 equal sub-areas.



**Solution**

In this problem,  $n = 3$  and  $D = 1.2$  m are given. The sampling points are on the circles with diameters of

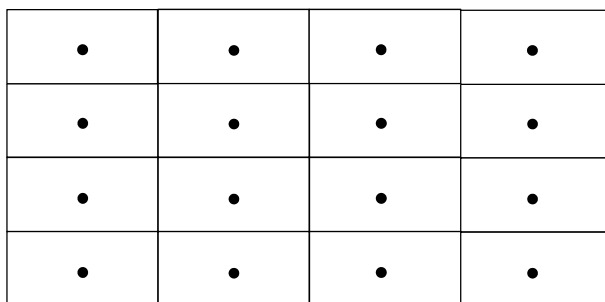
$$D_{si} = D\sqrt{i/2n} \quad \text{where } i = 1, 3 \text{ and } 5$$

$$D_{s1} = 1.2 \times \sqrt{1/6} = 0.4899 \text{ (m)}$$

$$D_{s2} = 1.2 \times \sqrt{3/6} = 0.8485 \text{ (m)}$$

$$D_{s3} = 1.2 \times \sqrt{5/6} = 1.0954 \text{ (m)}$$

These sampling points are depicted in Figure 8-8, where readers can determine the equal area circle  $D_i$  by using the same equation with  $i = 2, 4$  and  $6$ .



### Data Analysis

- The total gas flow rate discharged from a stack to the atmosphere is  $Q$  ( $\text{m}^3/\text{s}$ ),
- and a sample is taken with a volume flow rate of  $Q_s$  ( $\text{m}^3/\text{s}$ ) for a period of time,  $t$  (s).
- Analysis of this sample, either in situ or in a laboratory setting, shows that the mass of air pollutant of interest  $m$  (kg).
- With this information, one can calculate the mass concentration of the air pollutant in the sample ( $\text{kg}/\text{m}^3$ )

$$c = m/Q_s t$$

- If the sample is representative it can be used to represent the concentration of the air pollutant in the stack:  $C = c$ .
- Then the mass flow rate,  $\dot{m}$ , of this air pollutant in the stack is

$$\dot{m} = \frac{Qm}{Q_s t}$$

### Data Reporting

- A common standard for emission test report is to correct the concentrations against 50% excess air, 7% or 12% of  $\text{CO}_2$ , or 7% of  $\text{O}_2$ .
- Take 12% of  $\text{CO}_2$  as an example, the calculated emission rate is converted using the equation that follows. Actual procedure is much more complex, and readers are encouraged to seek local guidelines for more information.

$$C_{std (12\% CO_2)} = C_s \times \frac{12\%}{100\% \times x_{CO_2}}$$

$x_{CO_2}$  is the mole fraction of  $\text{CO}_2$  in the stack gas measured (e.g. following US EPA Method 3).